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85-039441/07 EISAI KK 25.07.83-JP-134260 (13.02.85) A61k-31/12 A61k-47 Aq. oleophilic vitamin soln. - is stabilised with hydrogenated lecithin and neutral amino acid	805 EISA 25.07.83 *EP -132-821-A	B(3-A, 3-H, 3-K, 4-B1B, 4-B2C, 5-B1P, 12-G1, 12-G7, 12-H4, 12-J1, 12-M6) 11 0 6 1
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<p>Hydrogenated lecithin and a neutral amino acid is added to an aq. soln. of a lipid-soluble active vitamin substance and/or ubiquinone to reduce side effects and stabilise. The pH is maintained at 5.5 to 8.</p> <p>Part of the water may be replaced by a water miscible solvent and an isotonsing substance may also be present.</p> <p><u>USE/ADVANTAGE</u></p> <p>Vitamin A is a growth promoter and activates visual function and reproduction and is a potential carcinostat. Vitamins E and K are biochemical antioxidant, biomembrane stabilisers and activators for blood clotting and electron transport. Ubiquinone is a cell activator, anti-oxidant and aldosterone antagonist.</p> <p>When the soln. is steam sterilised at 100°C for 1 hour the difference in transmittance at 640 nm before and after sterilisation is minimal.</p>		
<p><u>VITAMIN SUBSTANCE</u></p> <p>The vitamin substance includes vitamin A and its esters such as the palmitate, vitamin E and its esters such as the acetate or nicotinate and vitamins K<sub>1</sub>, K<sub>2</sub> and K<sub>3</sub> and their dihydro and demethyl derivs.</p> <p><u>HYDROGENATED LECITHIN</u></p> <p>The lecithin component includes hydrogenated ovollecithin but is pref. hydrogenated soybean lecithin. It should have an iodine value of 10 to 60, those with values lower than this do not disperse readily in aq. soln.</p> <p><u>NEUTRAL AMINOACID</u></p> <p>Aminoacids used include glycine, alanine, β-alanine, serine, threonine, valine, isoleucine, leucine, phenylalanine, methionine, histidine and taurine or their mixtures.</p> <p><u>ADDITIONAL SOLVENT</u></p> <p>Additional solvents include ethanol, propylene glycol, glycerol or low mol. wt. polyethylene glycol.</p> <p><u>ISOTONISING AGENT</u></p> <p>The isotonsing agent is e.g. glucose, xylitol, sorbitol or mannitol.</p> <p>EP-132821-A</p>		

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**EXAMPLE**

A soln. of vitamin K<sub>2</sub> (500 mg), hydrogenated soybean lecithin (500 mg) and propylene glycol (4g) and water was mixed ultrasonically under N<sub>2</sub> for 120 mins., then sorbitol (5g) and water (to 90 ml) were added. Proline (1g) was dissolved in the soln. and the pH adjusted to 7.0 with NaOH. Further water was added to make the soln. up to 100 ml. The soln. was filtered and sealed in brown ampoules and sterilised at 115°C for 30 mins. 23pp916LHDwgNo0/0)  
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⑤④ Aqueous solution containing lipid-soluble pharmaceutical substance and a process for preparing the same.

⑤⑦ Disclosed herein is an aqueous solution containing a lipid-soluble active vitamin substance and/or ubiquinone and a process for preparing the same. The solution has a pH of 5.5 - 8 and additionally contains a hydrogenated lecithin and neutral amino acid. It is stable and free from side effects, and is rather easy to prepare.

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AQUEOUS SOLUTION CONTAINING LIPID-SOLUBLE PHARMACEUTICAL  
SUBSTANCE AND A PROCESS FOR PREPARING THE SAME

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This invention relates to an improvement in or relating to an aqueous solution which contains a lipid-soluble pharmaceutical substance, namely, a lipid-soluble active vitamin substance and/or ubiquinone.

5 By the term "lipid-soluble active vitamin substance" is meant one or more substances selected from the group consisting of lipid-soluble active vitamin A substances, lipid-soluble active vitamin E substances and lipid-soluble active vitamin K substances.

10

Vitamin A has been known as a substance important for the promotion of growth, visual function and reproduction. It is recently attracting attention for its reported carcinostatic activities. On the other hand, vitamin E and  
15 vitamin K have been finding wide-spread clinical utility respectively owing to the biochemical anti-oxidation and biomembrane stabilization effects and as a substance pertaining to blood clotting and the electron transport system. These vitamins are these days desired to be  
20 available as aqueous solutions.

On the other hand, the term "ubiquinone" as used herein may embrace a variety of substances which individually contain different numbers of isoprene moieties in their structures. All of these ubiquinone substances have already  
25 been found that they are effective substances in view of

their physiological activities or effects such as supply of energy for cell activities, anti-oxidation effects, immune reinforcement reaction and aldosterone antagonism. Among such ubiquinone substances, a ubiquinone substance which is generally called ubidecarenone or CoQ<sub>10</sub> and contains 10 isoprene moieties has been being formed into pharmaceutical preparations.

Reflecting the finding of applicability of such pharmaceutical substances in a still broader field in recent years, a new desire has arisen for the availability of such pharmaceutical substances as aqueous solutions rather than as solid forms.

As a method for solubilizing the above-mentioned lipid-soluble active vitamin substance and/or ubiquinone, there is a conventional technique which makes use of a non-ionic surfactant, for example, HCO-60 (trade name; product of Nikko Chemical Co., Ltd., Japan). This prior art method however requires a great deal of HCO-60. As a result, the thus-prepared aqueous solution is susceptible of liberating histamine-like substances due to HCO-60 when administered as an injectable preparation, or when administered as an orally-dosable preparation, it develops troubles in the intestinal tract and thus brings about undesirable side effects such as diarrhea.

It has also been known to employ lecithin as an emulsifier. However, lecithin has weak emulsification capacity only. Therefore, this method requires a special apparatus called "pressure homogenizer". Moreover, the long-term stability of each resulting emulsion is not considered to be sufficient, thereby requiring such an

addit as vegetable oil or ethanol (see, Japanese Patent Application Laid-open No. 56315/1978).

Accordingly, the present inventors developed techniques both featuring an incorporation of a hydrogenated lecithin with a view toward providing an aqueous solution which contains a lipid-soluble active vitamin substance and/or ubiquinone and remains stable over a prolonged period of time without need for an addition of any additive having potential problems (see, Japanese Patent Application Nos. 209972/1981 and 212695/1982).

Thereafter, the present inventors made a further investigation on such aqueous solutions, resulting in a finding that it is preferable to adjust the pH of such an aqueous solution to 5.5 - 8. Since the aqueous solution is primarily used as an injectable preparation or orally-dosable preparation for pharmaceutical applications, it is desired to adjust its pH to the physiological pH range of living bodies, namely, to 5.5 - 8. Thus, additives were freely chosen to adjust the pH level of the above-mentioned aqueous solution. Such additives were incorporated to obtain final products. As a result, the resultant aqueous solutions tended to show some turbidity to eyes. When they were subjected to sterilization under suitable conditions, a significant clarity change was observed after the sterilization. Hence, it became necessary to make a search for an additive capable of adjusting the pH to 5.5 - 8 without developing any significant turbidity by such an adjustment, and a variety of substances was studied, leading to completion of the present invention.

With the foregoing in view, an object of this invention is to provide an aqueous solution containing a lipid-soluble active vitamin substance and/or ubiquinone which solution is free of the drawbacks of the above-described prior art techniques and has a pH adjusted to 5.5 - 8 and good stability without need for any additives having potential problems.

In one aspect of this invention, there is thus provided an aqueous solution containing a lipid-soluble active vitamin substance and/or ubiquinone, the improvement which comprises that the solution further comprises hydrogenated lecithin and a neutral amino acid incorporated therein and the pH of the solution ranges from 5.5 to 8.

The aqueous solution according to this invention has remarkable advantageous effects in that it is stable and does not induce side effects.

The above and other objects, features and advantages of the present invention will become apparent from the following description and the appended claims.

20

Briefly speaking, the present invention relates to an aqueous solution containing a lipid-soluble active vitamin substance and/or ubiquinone, which solution features an incorporation of a hydrogenated lecithin and neutral amino acid and a pH in the range of 5.5 - 8.

It is the principal feature of the aqueous solution according to this invention that when it is subjected to steam sterilization at a pH of 5.5 - 8, at 100°C and for 1 hour, its transmittance  $T_{640}$  is measured at 640 nm both before and after the sterilization and its percent change in

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transmittance (%) defined by the following equation is calculated, the percent change in transmittance is small.

$$\text{Percent change in transmittance (\%)} = \frac{T_{640}(\text{before sterilization}) - T_{640}(\text{after sterilization})}{T_{640}(\text{before sterilization})} \times 100$$

Therefore, the present invention provides an aqueous solution containing a lipid-soluble active vitamin substance and/or ubiquinone and having a small percent change in transmittance owing to a novel combination of constituent elements in this invention, which constituent elements will hereinafter be described specifically. The present invention will hereinafter be described more specifically.

As has already mentioned, the term "lipid-soluble active vitamin substance" as used herein means one or more substances selected from the group consisting of lipid-soluble active vitamin A substances, lipid-soluble active vitamin E substances and lipid-soluble active vitamin K substances.

As exemplary lipid-soluble active vitamin A substances useful in the practice of this invention, may be mentioned vitamin A per se and its esters, for example, vitamin A pulmitate. Illustrative of the lipid-soluble active vitamin E substance are vitamin E per se and its esters, for instance, vitamin E acetate and vitamin E nicotinate in the present invention. On the other hand, vitamins K<sub>1</sub>-K<sub>3</sub>, their dihydrogen derivatives and demethyl derivatives may be mentioned as exemplary lipid-soluble active vitamin K substances useful in the practice of this invention.

A variety of ubiquinones may be used in the present invention, among which ubidecarenone (CoQ<sub>10</sub>) is preferred.

On the other hand, the term "hydrogenated lecithin" as used herein means a lecithin the anti-oxidation property of

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which has been enhanced by its hydrogenation. More specifically, exemplary hydrogenated lecithins may embrace hydrogenated soybean lecithin; hydrogenated owolecithin and the like with hydrogenated soybean lecithin being particularly preferred. It is preferred that each of these hydrogenated lecithins contains at least 85% of a phospholipid component and has an iodine value of 10 - 60, notably 25 - 50. These iodine value ranges have been determined, because any iodine values smaller than 10 lead to lecithins, which have by themselves been hardened to considerable degrees and hence render their actual coarse dispersions difficult, whereas any iodine values in excess of 60 are not expected to bring about the advantageous effects of this invention. The hydrogenated lecithin may preferably contain, as the phospholipid component, phosphatidyl choline at a high level. In the case of a soybean phospholipid component for example, it may contain 80 - 95% of phosphatidyl choline. Besides, lysolecithin and phosphatidyl ethanolamine may also be detected. Particularly preferred hydrogenated lecithins are those recited in Japanese Patent Application Laid-open Nos. 83911/1977 and 62010/1980.

In the aqueous solution of this invention, the lipid-soluble active vitamin substance and/or ubiquinone and the hydrogenated lecithin may be contained in the following amounts.

First of all, the concentration of a lipid-soluble active vitamin and/or ubiquinone in an aqueous solution is required to be 0.1 - 1.0% from the clinical viewpoint. Concentrations in the range of 0.2 - 0.5% are popularly employed for usual applications. A concentration of 0.2% or so is often used especially when such aqueous solutions are

used as injectable preparations. It should however be borne in mind that the concentration of the lipid-soluble active vitamin and/or ubiquinone is not necessarily limited to the above range.

5           The hydrogenated lecithin may be incorporated at various different concentration levels in accordance with what end use would be made on the resulting aqueous solutions. When an aqueous solution is desired to be clear in view of its application purpose, it is preferred to add  
10           the hydrogenated lecithin in an amount of 1 - 5 parts by weight per part by weight of the lipid-soluble active vitamin substance and/or ubiquinone. Where slight cloudiness is tolerated for an aqueous solution, the hydrogenated lecithin may be incorporated from as little as 0.1 - 1 parts by  
15           weight to as much as 5 - 15 parts by weight, both, per part by weight of the lipid-soluble active vitamin substance and/or ubiquinone. For practical purposes, the hydrogenated lecithin may be added in an amount of 0.1 - 15 parts by  
20           weight per part by weight of the lipid-soluble active vitamin substance and/or ubiquinone. However, it is not necessary to limit the concentration of the hydrogenated lecithin to the above-mentioned ranges in the present invention.

          In the aqueous solution of this invention, a part of  
25           water may be replaced by a water-miscible solvent such as ethanol, propylene glycol, a low molecular weight polyethylene glycol, glycerin or the like. These solvents are effective in shortening to a considerable extent the time required to disperse coarsely and evenly the  
30           lipid-soluble active vitamin substance and/or ubiquinone in the first step upon preparation of the aqueous solution. Namely,

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the time required for the solubilization of the lipid-soluble active vitamin substance and/or ubiquinone may be shortened when the lipid-soluble active vitamin substance and/or ubiquinone is in advance dispersed coarsely in the

5 water-miscible solvent in the presence of the hydrogenated lecithin instead of mixing the lipid-soluble active vitamin substance and/or ubiquinone directly with the hydrogenated lecithin and then adding water to the resultant mixture. It should however be borne in mind that the water-miscible

10 solvent may be used to facilitate the preparation of the aqueous solution of this invention and the object of this invention can thus still be attained without incorporation of such a water-miscible solvent.

Therefore, it is not essential for the present

15 invention to add such a solvent.

If the water-miscible solvent is added in order to facilitate the preparation of the aqueous solution of this invention, it may be incorporated in an amount of 1 - 50 parts by weight per part by weight of the lipid-soluble

20 active vitamin substance and/or ubiquinone and in an amount of 2 - 10 wt./vol.% of the aqueous solution of this invention.

When the aqueous solution of this invention is used as an injectable preparation, it is possible to add a sugar and/or sugar alcohol such as glucose, xylitol, sorbitol,

25 mannitol and/or the like, which are commonly and widely used as isotonizing agents in injectable preparations. Namely, addition of these isotonizing agents are effective in avoiding occurrence of a haze or the like upon sterilization of injectable preparations without deleteriously affecting

30 the meritorious features of this invention. It is preferred

to add such an isotonizing agent in an amount of 1 - 10% of the aqueous solution of this invention.

Next, the term "neutral amino acid" as used herein means an amino-containing acid, an aqueous solution of which has a pH in the neutral range. As its specific and representative examples, there may be mentioned glycine, alanine,  $\beta$ -alanine, serine, threonine, valine, isoleucine, leucine, phenylalanine, methionine, histidine and taurine. These amino acids may be used either singly or in combination.

Since the pH of an aqueous solution of each of these neutral amino acids falls within the neutral range, the neutral amino acid can adjust the pH of the aqueous solution of this invention to 5.5 to 8. In some instances, they may show buffer actions to the pH level. It has however been unknown to date that at the same time, they serve to keep the percent change in transmittance of the aqueous solution of this invention at a small value.

It is preferred to limit the concentration of such a neutral amino acid to 0.05 - 6 wt./vol.% in the aqueous solution of this invention. Needless to say, it is essential to limit the concentration of the neutral amino acid to the above range.

The pH of the aqueous solution of this invention is limited to 5.5 - 8. This limitation to the pH range is an essential requirement for the aqueous solution of this invention in view of its physiological application. Such a pH range can be achieved by incorporating one or more of the neutral amino acids and if necessary, adding an acidic or alkaline substance to make a fine pH adjustment.

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The aqueous solution of this invention may be prepared in a manner to be outlined hereinbelow. First of all, a small amount of water is added to the lipid-soluble active vitamin substance and/or ubiquinone and the hydrogenated lecithin. Thereafter, the resulting mixture is coarsely and evenly dispersed preferably while heating it at 60 - 70°C. For the sake of efficient dispersion, it is preferred to apply a pressure or ultrasonic waves while agitating the mixture, so that the mixture is forced to disperse. The coarse dispersion may be facilitated further when a water-miscible solvent such as ethanol, propylene glycol, a low molecular weight polyethylene glycol, glycerin or the like is used in lieu of a part of water. Then, the neutral amino acid, other components and the remaining water are added to the thus-obtained coarse dispersion and the resulting mixture is dispersed evenly, leading to the provision of the aqueous solution of this invention. When using the aqueous solution of this invention as an injectable preparation, it is necessary to filter the aqueous solution, fill the thus-filtered aqueous solution in desired ampules and sterilize it in the ampules. Incidentally, a pharmaceutically-acceptable germicide, isotonizing agent and/or the like may also be chosen at will as other materials to be incorporated. However, it is preferable to avoid addition of any electrolytic component because it hinders the dispersion, especially the solubilization.

By the way, the hydrogenated lecithin may be prepared by charging lecithin in an autoclave, adding a solvent and catalyst into the autoclave, and then causing the hydrogenation of the lecithin to proceed to a desired iodine

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value while maintaining the reaction mixture in contact with hydrogen. After completion of the reaction, the catalyst is filtered off and the solvent is then distilled off to obtain the hydrogenated lecithin.

As the hydrogenated lecithin useful in the practice of this invention, it is particularly preferred to use a specially-purified hydrogenated lecithin such as that disclosed in Japanese Patent Application Laid-open No. 62010/1980 which was referred to in the above.

The present invention will hereinafter be described in further detail in the following Examples, which will be given by way of example only and not by way of limitation of the invention.

Example 1:

Added to vitamin K<sub>2</sub> (500 mg) were hydrogenated soybean lecithin (500 mg), propylene glycol (4 g) and water (20 ml). The resulting mixture was stirred with heating and was then subjected to an ultrasonic processing (20 KHz, 200W) for 120 minutes in an nitrogen-substituted atmosphere to obtain an aqueous solution. Sorbitol (5 g) and the remaining water were added to the thus-prepared aqueous solution to make the total volume be 90 ml, followed by dissolution of proline (1 g). The pH of the resulting aqueous mixture was adjusted to 7.0 with an aqueous solution of sodium hydroxide. Then, water was added in such an amount that the total volume of the resulting mixture was increased to 100 ml. The thus-obtained aqueous solution was filtered by a membrane filter, purged with nitrogen gas, and filled in brown ampules. The ampules were melt-sealed and were then sterilized at 115°C for 30 minutes to obtain stable injectable preparations which contained vitamin K<sub>2</sub>.

Example 2:

Hydrogenated and purified ovolcithin (250 mg),  
glycerin (3 g) and water (30 mg) were added to vitamin E  
nicotinate (200 mg). The resulting mixture was agitated  
5 with heating, and was then stirred for 50 minutes in a  
nitrogen-substituted atmosphere by a high-speed stirrer to  
obtain an aqueous solution. Then, sorbitol (20 g), methyl-  
paraben (100 mg), orange essence and glycine (500 mg) were  
added to the aqueous solution, followed by an adjustment of  
10 the pH of the resulting mixture to 7.0 with an aqueous  
solution of sodium hydroxide. Thus, a stable syrup  
containing vitamin E nicotinate (100 ml in total) was  
obtained.

No significant change was observed in the transmittance  
15 ( $T_{640}$ ) of the syrup even after it had been allowed to  
stand at room temperature for 1 month.

Example 3:

Added to free vitamin E (500 mg), were hydrogenated and  
purified soybean lecithin (2 g), purified sesame oil (500  
20 mg), propylene glycol (6g) as a solubilizing additive and  
water (30 ml). The resulting mixture was stirred with  
heating, and was then stirred in a nitrogen-substituted  
atmosphere for 90 minutes by a high-speed stirrer to obtain  
an aqueous solution. Sorbitol (5 g) and serine (500 mg)  
25 were added to the aqueous solution. The pH and total volume  
of the resulting mixture were adjusted respectively to 6.5  
and 100 ml with an aqueous solution of sodium hydroxide.  
Thereafter, the procedures of Example 1 were repeated in  
much the same way to obtain a stable injectable preparation  
30 which contained free vitamin E.

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Example 4.

Hydrogenated and purified soybean lecithin (300 mg), propylene glycol (3 g) as a solubilizing additive and water (30 mg) were added to vitamin A pulmitate (200 mg). The resulting mixture was stirred with heating. It was processed for 60 minutes in a nitrogen-substituted atmosphere by means of a high-speed stirrer to obtain an aqueous solution. Then, xylitol (3 g), ethanol (10 g) and serine (300 mg) were added to the aqueous solution. The pH and total volume of the resulting mixture were adjusted respectively to 6.5 and 100 ml with an aqueous solution of sodium hydroxide to obtain a stable solution suitable for endermic applications and containing vitamin A pulmitate.

No significant change was observed in the transmittance ( $T_{649}$ ) of the solution even after it had been allowed to stand at room temperature for 1 month.

Example 5:

Water (4 ml) was added to a mixture of ubidecarenone (300 mg), hydrogenated and purified soybean lecithin (400 mg) and sorbitol (4 g). In a nitrogen-substituted atmosphere, the resulting mixture was subjected for 5 minutes to an ultrasonic processing, followed by further addition of water (20 ml) and propylene glycol (4 g). The ultrasonic processing was carried out for 10 minutes to obtain a clear aqueous solution. The rest of the water was added to the aqueous solution to increase its total volume to 90 ml, followed by an addition of alanine (300 mg). The pH of the resulting mixture was adjusted to 6.5 with an aqueous solution of sodium hydroxide. Water was then added to the resulting solution to make its total volume be 100 ml. Thereafter, the procedures of Example 1 were followed

in much the same way to obtain a stable injectable preparation which contained ubidecarenone.

The advantageous effects of this invention will hereinafter be described in the following Experiments, in comparison with some comparative examples.

Experiment 1:

Hydrogenated and purified soybean lecithin (220 mg), propylene glycol (3 g) as a solubilizing additive and water (20 ml) were added to ubidecarenone (250 mg). The resulting mixture was stirred with heating. Then, it was subjected for 90 minutes to an ultrasonic processing (20 KHz, 200 W) in a nitrogen-substituted atmosphere to obtain an aqueous solution. Then, sorbitol (5 g) and the remaining water were added to the resulting mixture to make its total volume be 90 ml. Using the various pH-adjusting agents given in Table 1, aqueous solutions obtained in the above-described manner were respectively subjected to pH-adjustment. The volumes of the resulting aqueous solutions were increased to the prescribed level to obtain aqueous solutions each of which contained 0.25% of ubidecarenone. Each of the resulting sample solutions was filtered through a membrane filter, purged with nitrogen gas, and filled in 2-ml portions in ampules. The ampules were melt-sealed and were then steam-sterilized at 100°C for 1 hour. Their pH values and transmittance values were measured both before and after the sterilization. Changes in transmittance were calculated on the basis of the transmittance values. Results are summarized in Table 1.

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Table 1

pH-Adjusting agent		Before sterilization		After sterilization		Percent change(%)
Name	Conc. (%)	pH	T <sub>640</sub> (%)	pH	T <sub>640</sub> (%)	
Not added	-	6.09	95.2	5.38	94.0	1.3
Aspartic acid	0.1	7.49	91.0	7.03	61.2	32.7
Arginic acid	0.1	7.48	90.8	7.53	55.5	38.9
Megrumic acid	0.1	7.48	90.2	7.14	75.0	16.9
Citric acid	0.1	7.48	79.5	7.52	37.0	53.5
Glutamic acid	0.1	7.47	90.6	6.93	64.6	28.7
Glycine	0.1	7.51	93.5	7.04	94.5	0
ditto	0.1	7.07	94.4	6.64	94.0	0.4
ditto	0.1	6.60	94.8	6.47	94.3	0.5
ditto	0.5	7.51	93.0	7.24	92.8	0.2
ditto	0.5	7.00	93.5	6.76	93.2	0.3
ditto	0.5	6.58	94.2	6.50	93.8	0.4
Alanine	0.1	7.53	95.0	7.01	94.4	0.6
ditto	0.1	7.08	95.0	6.68	93.5	1.6
ditto	0.1	6.58	95.2	6.57	94.0	1.3
ditto	0.5	7.49	93.5	7.27	92.6	1.0
ditto	0.5	7.00	94.0	6.75	93.5	0.5
ditto	0.5	6.53	94.0	6.45	93.4	0.6

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Table 1 (Cont'd)

pH-Adjusting agent		Before sterilization		After sterilization		Percent change(%)
Name	Conc. (%)	pH	T <sub>640</sub> (%)	pH	T <sub>640</sub> (%)	
Threonine	0.1	7.52	94.2	6.93	94.2	0
ditto	0.1	7.02	94.8	6.44	95.0	0
ditto	0.1	6.51	95.0	6.42	95.3	0
ditto	0.5	7.51	92.7	7.02	92.8	0
ditto	0.5	7.00	93.4	6.59	94.0	0
ditto	0.5	6.50	94.8	6.19	94.3	0.5
Serine	0.1	7.49	94.3	6.85	94.6	0
ditto	0.1	6.97	94.8	6.47	94.8	0
ditto	0.1	6.50	94.7	6.28	95.0	0
ditto	0.5	7.50	92.3	7.17	92.7	0
ditto	0.5	7.00	93.6	6.62	94.4	0
ditto	0.5	6.51	93.7	6.31	93.8	0
Proline	0.1	7.72	93.5	7.01	93.0	0.5
Valine	0.1	7.77	92.0	7.42	92.5	0
Isoleucine	0.1	7.86	91.5	7.58	92.0	0
Methionine	0.1	7.78	91.0	7.63	92.0	0
Phenylalanine	0.1	7.48	93.5	7.49	92.5	1.1
Histidine	0.1	7.48	95.0	7.55	93.0	2.1
Taurine	0.1	7.79	91.0	7.68	91.0	0

As apparent from Table 1, the preparation added with no pH-adjusting agent had good apparent stability but its pH was lowered by the sterilization to a level outside the stable range for lecithin. Furthermore, the preparations, the pH levels of which were adjusted with the acidic or basic amino acids, developed considerable changes in appearance.

Turning on the other hand to the preparations, the pH levels of which were respectively effected with the neutral amino acids in accordance with this invention, their pH levels ranged from 5.5 to 8 and no substantial changes were observed with respect to their appearance.

Experiment 2:

To either one (250 mg) of ubidecarenone, vitamin E acetate and vitamin K<sub>1</sub>, were added with heating hydrogenated and purified ovollecithin (220 mg), propylene glycol (5 g) as a solubilizing additive and water (20 ml). The resulting mixture was subjected to an ultrasonic processing for 90 minutes in a nitrogen-substituted atmosphere. Thereafter, its pH was adjusted in much the same way as Example 1 to obtain a sample. The sample was subjected at 100°C for 1 hour to steam sterilization. Its pH values and transmittance values were measured respectively both before and after the sterilization. A change in transmittance was then calculated on the basis of the transmittance values. Results are given in Table 2.

Table 2

Lipid-soluble pharmaceutical substance	pH-Adjusting agent	Before sterilization		After sterilization		Percent change (%)
		pH	Transmittance (%)	pH	Transmittance (%)	
Ubidecarenone	Not added	4.17	89.3	4.34	87.8	1.7
	Phosphoric acid 0.1%	7.56	87.5	7.42	78.2	10.6
	Glycine 0.1%	7.54	88.3	7.22	86.5	2.0
Vitamin E acetate	Not added	4.38	87.2	4.49	85.5	1.9
	Glutamic acid 0.1%	7.51	86.0	6.78	75.2	12.6
	Serine 0.1%	7.51	86.5	7.34	84.9	1.8
Vitamin K <sub>1</sub>	Not added	4.46	88.9	4.52	85.9	3.4
	Aspartic acid 0.1%	7.51	85.9	6.79	76.8	10.6
	Alanine 0.1%	7.53	87.8	6.83	84.2	4.1

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As clearly envisaged from Table 2 and similar to the preparations making use of hydrogenated and purified soybean lecithin, the preparations the pH levels of which were adjusted respectively with the neutral amino acids in accordance with this invention developed less changes compared with those pH-adjusted using phosphoric acid and the acidic amino acids respectively and were thus aqueous solutions more stable than the latter preparations.

Having now fully described the invention, it will be apparent to one of ordinary skill in the art that many changes and modifications can be made thereto without departing from the spirit or scope of the invention as set forth herein.

## CLAIMS:

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1. In an aqueous solution containing a lipid-soluble active vitamin substance and/or ubiquinone, the improvement which comprises that the solution further comprises hydrogenated lecithin and a neutral amino acid incorporated  
5 therein and the pH of the solution ranges from 5.5 to 8.

2. An aqueous solution according to Claim 1, wherein the hydrogenated lecithin is contained in an amount of 0.1 - 15 parts by weight per part by weight of the lipid-soluble  
10 active vitamin substance and/or ubiquinone.

3. An aqueous solution according to Claim 1, wherein the hydrogenated lecithin contains at least 85% of a phospholipid component and has an iodine value of 10 - 60.  
15

4. An aqueous solution according to Claim 3, wherein the hydrogenated lecithin is hydrogenated soybean lecithin or hydrogenated ovollecithin.

20 5. An aqueous solution according to Claim 1, wherein the neutral amino acid is contained in an amount of 0.05 - 6 wt./vol.% of the aqueous solution.

25 6. An aqueous solution according to Claim 5, wherein the neutral amino acid is one or more neutral amino acids selected from the group consisting of glycine, alanine,  $\beta$ -alanine, serine, threonine, valine, isoleucine, leucine, phenylalanine, methionine, histidine and taurine.

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7. An aqueous solution according to Claim 1, wherein the aqueous solution additionally contains a water-miscible solvent and/or an isotonizing agent.

8. An aqueous solution according to Claim 7, wherein the water-miscible solvent is ethanol, propylene glycol, a low molecular weight polyethylene glycol or glycerin.

10 9. An aqueous solution according to Claim 7, wherein the water-miscible solvent is contained in an amount of 1 - 50 parts by weight per part by weight of the lipid-soluble active vitamin substance and/or ubiquinone and in an amount of 2 - 10% of the aqueous solution.

15 10. An aqueous solution according to Claim 7, wherein the isotonizing agent is a sugar or sugar alcohol.

20 11. An aqueous solution according to Claim 10, wherein the isotonizing agent is glucose, xylitol, sorbitol or mannitol.

12. An aqueous solution according to Claim 10, wherein the isotonizing agent is contained in an amount of 0.05 - 6 wt./vol.% of the aqueous solution.

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13. A process for the preparation of an aqueous solution containing lipid-soluble pharmaceutical substance, which comprises adding, to water, (i) the lipid-soluble active vitamin substance and/or ubiquinone, and (ii) a hydrogenated lecithin, dispersing the resulting mixture, followed by adding a neutral amino acid.

14. A process as claimed in Claim 13, wherein a part of the water is replaced with a water-miscible solvent.